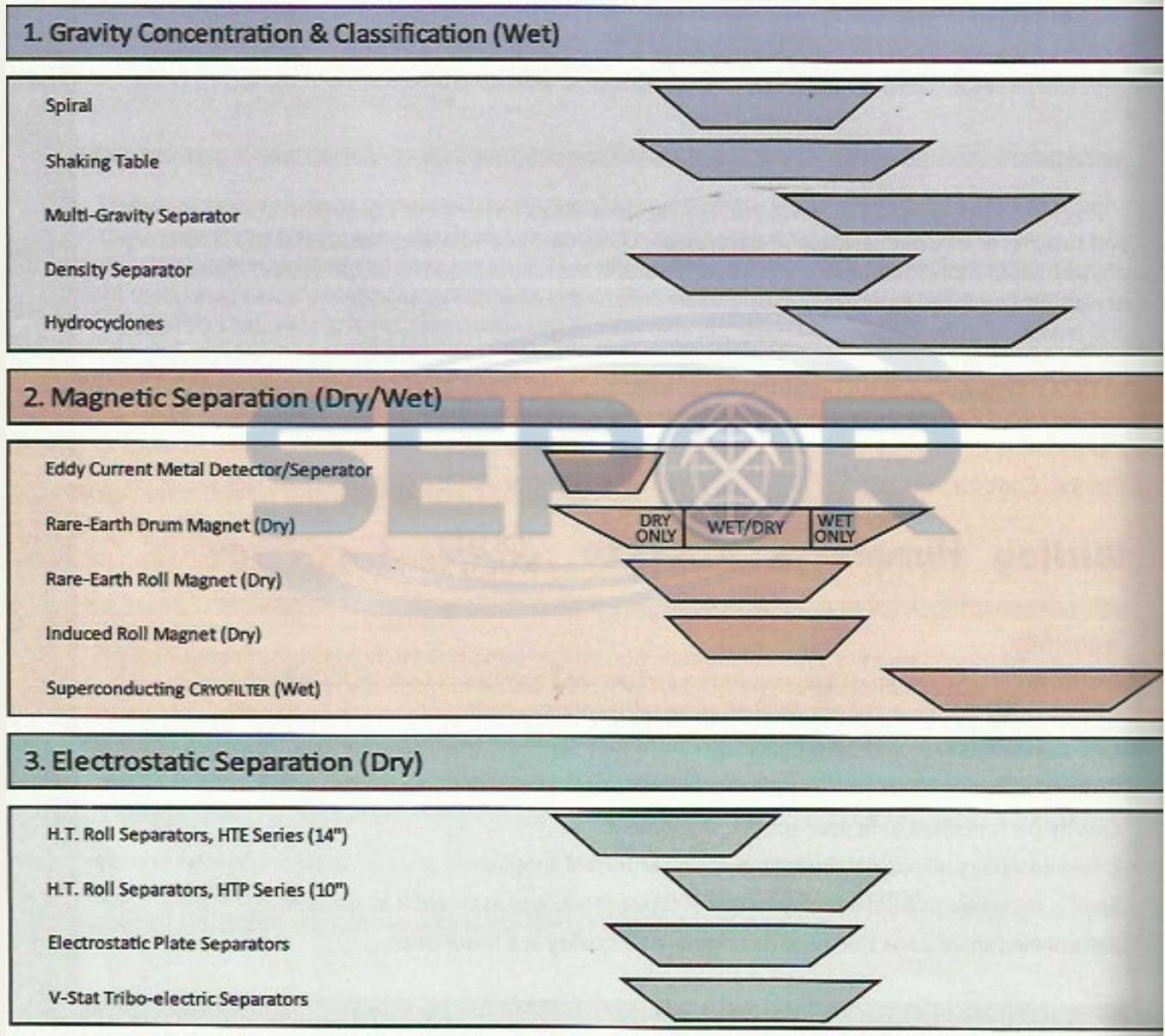


# Magnetic & Electrostatic Separation Concentration Method vs Size Range

10 mm 1.0 mm 0.1 mm 0.075 mm 0.045 mm 0.001 mm



# Magnetic & Electrostatic Separation Response Chart

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Physical Characteristics of Select Minerals and Materials			Magnetic Response			Electrostatic Response	
Mineral	Composition	Specific Gravity	F	P	NM	C	NC
Actinolite	$\text{Ca}_2(\text{Mg,Fe})_5[\text{Si}_4\text{O}_{11}]_2(\text{OH})_2$	3.0-3.2		X			X
Albite	$\text{Na}[\text{AlSi}_3\text{O}_8]$	2.6			X		X
Almandine	$\text{Fe}_3\text{Al}_2[\text{SiO}_4]_3$	4.3		X			X
Amphibole	$(\text{Fe,Mg,Ca})_2\text{SiO}_2$	2.9-3.5		X			X
Anatase	$\text{TiO}_2$	3.9			X	X	
Andalusite	$\text{Al}_2\text{SiO}_5$	3.2			X		X
Andradite	$3\text{CaO}\cdot\text{Fe}_2\text{O}_3\cdot 3\text{SiO}_2$	3.8		X		(2)	X
Anhydrite	$\text{CaSO}_4$	3			X		X
Ankerite	$\text{Ca}(\text{Mg,Fe})[\text{CO}_3]_2$	2.9-3.1		X			X
Apatite	$(\text{F,Cl,OH})\text{Ca}_5(\text{PO}_4)_3$	3.2			X		X
Aragonite	$\text{CaCO}_3$	3			X		X
Arsenopyrite	$\text{FeAsS}$	5.9-6.2		X → (1)		X	
Asbestos	$\text{Mg}_3[\text{Si}_2\text{O}_5](\text{OH})_4$	2.4-2.5			X		X
Augite	$\text{Ca}(\text{Mg,Fe,Al})_2[\text{Si,Al}_2\text{O}_6]$	3.2-3.5		X		X	(1)
Azurite	$\text{Cu}_2[\text{CO}_3]_2(\text{OH})_2$	3.8			X		X
Baddeleyite	$\text{ZrO}_2$	5.6			X		X
Barite	$\text{BaSO}_4$	4.5			X		X
Bastnaesite	$(\text{Ce,La,F})\text{CO}_3$	5		X			X
Bauxite	$\text{Al}_2\text{O}_3\cdot 2\text{H}_2\text{O}$	2.6			X		X
Beryl	$\text{Be}_3\text{Al}_2[\text{Si}_4\text{O}_{12}]$	2.7-2.8			X		X
Biotite	$\text{K}(\text{Mg,Fe})_2[\text{Si}_4\text{AlO}_{10}](\text{OH,F})_2$	3.0-3.1		X			(4)
Bismuth	Bi	9.8			X	X	
Borax	$\text{Na}_2\text{B}_4\text{O}_7\cdot 10\text{H}_2\text{O}$	1.7			X		X
Bornite	$\text{Cu}_5\text{FeS}_4$	4.9-5.0		(1) ← X	X	X	
Brannerite	$(\text{UO,TiO,UO}_2)\text{TiO}_3$	4.5-5.4		X		X	
Brookite	$\text{TiO}_2$	4.1			X	X	
Calcite	$\text{CaCO}_3$	2.7			X		X
Carotite	$\text{K}_2(\text{UO}_2)_2\text{V}_2\text{O}_8\cdot 2\text{H}_2\text{O}$	5			X	(2)	X
Cassiterite	$\text{SnO}_2$	7			X	X	
Celestite	$\text{SrSO}_4$	4			X		X
Cerussite	$\text{PbCO}_3$	6.6			X	(2)	X
Chalcoite	$\text{Cu}_2\text{S}$	5.5-5.8			X	X	
Chalcopyrite	$\text{CuFeS}_2$	4.1-4.3		(1) ← X	X	X	
Chlorite	$(\text{Mg,Al,Fe})_2[\text{Si,Al}_x\text{O}_{10}](\text{OH})_{2-x}$	2.6-3.2		X			X
Chromite	$(\text{Fe,Mg})(\text{Cr,Al})_2\text{O}_4$	4.6		X		X	
Chrysocolla	$\text{CuSiO}_3\cdot 2\text{H}_2\text{O}$	2.0-2.3			X		X
Cinnabar	HgS	8.1			X		X
Cobalite	$(\text{Co,Fe})\text{AsS}$	6.0-6.3		X		X	
Colemanite	$\text{Ca}_2\text{B}_4\text{O}_{11}\cdot 5\text{H}_2\text{O}$	2.4			X		X
Collpohanite	$\text{Ca}_3\text{P}_2\text{O}_7\cdot \text{H}_2\text{O}$	2.6-2.9			X		(3)
Columbite	$(\text{Fe,Mn})(\text{Ta,Nb})_2\text{O}_6$	5.2-8.2		X		X	
Copper	Cu	8.9			X	X	
Corundum	$\text{Al}_2\text{O}_3$	3.9-4.1			X		X
Covellite	$\text{CuS}$	4.7			X	X	
Cryolite	$\text{Na}_3\text{AlF}_6$	3			X	(2)	X
Cuprite	$\text{Cu}_2\text{O}$	5.8-6.2			X		X
Diamond (natural)	C	3.5			X		X
Diamond (synthetic)	C	3.5		X			X
Diopside	$\text{CaMg}[\text{Si}_2\text{O}_6]$	3.3-3.4		X → (1)			X
Dolomite	$\text{CaMg}(\text{CO}_3)_2$	1.8-2.9			X		X

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Mineral	Composition	Specific Gravity	F	P	NM	C	NC
Epidote	$\text{Ca}_2(\text{AlFe})_2\text{Si}_2\text{O}_7(\text{OH})$	3.4		X			X
Euxenite	$(\text{Y}, \text{Er}, \text{Ce}, \text{La}, \text{U})(\text{Nb}, \text{Ti}, \text{Ta})_2$	4.7-5.2		X		X	
Feldspar Group	$(\text{K}, \text{Na}, \text{Ca} \dots)(\text{Al}, \text{Si})_3\text{O}_8$	2.6-2.8			X		X
Ferberite	$\text{FeWO}_4$	7.5	(1) ←	X		X	
Flint	$\text{SiO}_2$	2.6			X		X
Flourite	$\text{CaF}_2$	3.2			X		X
Franklinite	$(\text{Zn}, \text{Mn})\text{Fe}_2\text{O}_4$	5.1-5.2	X			X	
Gahnite	$\text{ZnAl}_2\text{O}_4$	4.6			X		X
Galena	$\text{PbS}$	7.5			X	X	
Garnet	Complex Ca, Mg, Fe, Mn Silicates	3.4-4.3		X → (1)		(2)	X
Gibbsite	$\text{Al}(\text{OH})_3$	2.4			X		X
Goethite	$\text{HFeO}_2$	4.3		X		(2)	X
Gold	$\text{Au}$	19.3			X	X	
Graphite	$\text{C}$	2.1-2.2			X	X	
Grossularite	$\text{Ca}_3\text{Al}_2(\text{SiO}_4)_3$	3.5			X	(2)	X
Gypsum	$\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$	2.3			X		X
Halite	$\text{NaCl}$	2.5			X	(2)	X
Hematite	$\text{Fe}_2\text{O}_3$	5.2		X		X	
Hornblende	$\text{Ca}_2\text{Na}(\text{Mg}, \text{Fe}^{2+})_4(\text{Al}, \text{Fe}^{3+})$	3.1-3.3		X		(2)	X
Huebnerite	$\text{MnWO}_4$	6.7-7.5		X → (1)		X	
Hypersthene	$(\text{Mg}, \text{Fe})\text{SiO}_3$	3.4		X			X
Ilmenite	$\text{FeTiO}_3$	4.7		X		X	
Ilmenorutile	$(\text{Nb}_2\text{O}_5, \text{Ta}_2\text{O}_5)_x\text{TiO}_2$	5.1		X		X	
Ilvaite	$\text{CaFe}_2(\text{FeOH})(\text{SiO}_4)_3$	4		X		X	(1)
Kaolinite	$\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$	2.6			X		X
Kyanite	$\text{Al}_2\text{O}_3(\text{SiO}_4)$	3.6-3.7			X		X
Lepidolite	$(\text{OH}, \text{F})_2\text{KLiAl}_3\text{Si}_3\text{O}_{10}$	2.8-2.9			X		X
Leucokene	$\text{FeTiO}_3 \rightarrow \text{TiO}_2$ (Alteration Product)	3.6-4.3		X → (1)		X	
Limonite	$\text{HFeO}_2 \cdot n\text{H}_2\text{O}$	2.2-2.4		X → (1)		(2)	X
Magnesite	$\text{MgCO}_3$	3			X		X
Magnetite	$\text{Fe}_3\text{O}_4$	5.2	X			X	
Malachite	$\text{Cu}_2\text{CO}_3(\text{OH})_2$	4			X		X
Manganite	$\text{MnO}(\text{OH})$	4.2		X → (1)		X	
Marcasite	$\text{FeS}_2$	4.6-4.9		(1) ←	X	X	
Martite	(See Hematite)						
Microlite	$\text{KAlSi}_4\text{O}_{12}$	2.6			X		X
Microlite	$\text{Ca}_2\text{Ta}_2\text{O}_7$ (See Pyrochlore)	5.5			X		X
Millerite	$\text{NiS}$	5.2-5.6		X		X	
Molybdenite	$\text{MoS}_2$	4.7-5.0			X	X	
Monazite	$(\text{Ce}, \text{La}, \text{Y}, \text{Th})\text{PO}_4$	4.9-5.5		X			X
Mullite	$\text{Al}_2\text{Si}_2\text{O}_7$	3.2			X		X
Muscovite	$\text{KA}_{2-3}(\text{AlSi}_2\text{O}_6)(\text{F}, \text{OH})_2$	2.9-3.0			X		(4)
Nahcolite	$\text{NaHCO}_3$	2.2			X		X
Nepheline Syenite	$(\text{Na}, \text{K})(\text{AlSi})_3\text{O}_8$	2.6			X		X
Nicolite	$\text{NiAs}$	7.6-7.8		X		X	
Olivine	$(\text{Mg}, \text{Fe})_2(\text{SiO}_4)$	3.3-3.5		X			X
Orpiment	$\text{As}_2\text{S}_3$	3.4-3.5			X	X	
Orthoclase	$\text{K}(\text{AlSi}_3\text{O}_8)$	2.5-2.6			X		X
Periclase	$\text{MgO}$	3.6			X		X
Perovskite	$\text{CaTiO}_3$	4			X		X
Petalite	$\text{Li}(\text{AlSi}_4\text{O}_{12})$	2.4			X		X
Phosphate (pebble)	(See Collaphanite)						
Platinum	$\text{Pt}$	14.0-21.5		(1) ←	X	X	
Pyrite	$\text{FeS}_2$	5		(1) ←	X	X	
Pyrochlore	$(\text{Na}, \text{Ca} \dots)_2(\text{Nb}, \text{Ta} \dots)_2\text{O}_6(\text{F}, \text{OH})$	4.2-4.4			X	X	
Pyrolusite	$\text{MnO}_2$	4.7-5.0		(1) ←	X		X
Pyrope	$\text{Mg}_3\text{Al}_2(\text{SiO}_4)_3$	3.5			X	(2) ←	X
Pyroxene	$(\text{Ca}, \text{Mg}, \text{Fe}, \text{Al})_2\text{SiO}_6$	3.1-3.6		X → (1)		(2) ←	X
Pyrrhotite	$\text{Fe}_{1-x}\text{S}_x$	4.6-4.7	X			X	

Physical Characteristics of Select Minerals and Materials			Magnetic Response			Electrostatic Response	
Mineral	Composition	Specific Gravity	F	P	NM	C	NC
Quartz	SiO <sub>2</sub>	2.7			X		(3)
Realgar	As <sub>2</sub> S <sub>3</sub>	3.6			X	X	
Rhodochrosite	MnCO <sub>3</sub>	3.7			X	(2) ←	X
Rhodonite	MnSiO <sub>3</sub>	3.6-3.7			X	(2) ←	X
Rutile	TiO <sub>2</sub>	4.2-4.3			X	(2)	
Samaraskite	(Y,Er...) <sub>4</sub> [Nb,Ta] <sub>2</sub> O <sub>7</sub>	5.6-5.8	(1) ←	X		X	
Scheelite	CaWO <sub>4</sub>	6.1			X		X
Serpentine	Mg <sub>3</sub> [Si <sub>4</sub> O <sub>10</sub> ](OH) <sub>2</sub>	2.5-2.7		X			X
Siderite	FeCO <sub>3</sub>	3.9		X		(2) ←	X
Sillimanite	Al <sub>2</sub> O <sub>3</sub> [SiO <sub>4</sub> ]	3.2			X		X
Silver	Ag	10.1-11.1			X	X	
Smithsonite	ZnCO <sub>3</sub>	4.1-4.5			X		X
Sodalite	Na <sub>4</sub> [Al <sub>6</sub> Si <sub>6</sub> O <sub>24</sub> ]Cl <sub>2</sub>	2.1-2.3			X		X
Spessartite	Mn <sub>2</sub> Al <sub>2</sub> (SiO <sub>4</sub> ) <sub>2</sub>	4.2			X		X
Sphalerite	ZnS	3.9-4.0		X →	(1)	X →	(1)
Sphene	CaTi(SiO <sub>4</sub> )(F,OH)	3.3-3.6			X	(2) ←	X
Spinel	MgAl <sub>2</sub> O <sub>4</sub>	3.6		(1) ←	X	X	(1)
Spodumene	LiAl(SiO <sub>3</sub> ) <sub>2</sub>	3.1-3.2			X		X
Stannite	Cu <sub>2</sub> FeSnS <sub>4</sub>	4.3-4.5			X	X	
Staurolite	Fe <sup>2+</sup> Al <sup>2+</sup> [Si <sup>4+</sup> O <sub>11</sub> ] <sub>2</sub> O <sub>2</sub> (OH) <sub>2</sub>	3.6-3.8		X		(2) ←	X
Stibnite (Antimonite)	Sb <sub>2</sub> S <sub>3</sub>	4.6			X	X	
Struverite	(Ta <sub>2</sub> O <sub>5</sub> ,Nb <sub>2</sub> O <sub>5</sub> ) <sub>2</sub> TiO <sub>2</sub>	5.1		X		X	
Sulphur	S	2.1			X		X
Sylvite	KCl	2			X		X
Talc	Mg <sub>3</sub> Si <sub>4</sub> O <sub>10</sub> (OH) <sub>2</sub>	2.7-2.8			X		X
Tantalite	(FeMn)(TaNb) <sub>2</sub> O <sub>6</sub>	5.2-8.2		X		X	
Tapiolite	Fe(Ta,Nb) <sub>2</sub> O <sub>6</sub>	7.3-7.8		X		X	
Tetrahedrite	(Cu,Fe) <sub>12</sub> Sb <sub>4</sub> S <sub>14</sub>	5		X		X	
Thorianite	ThO <sub>2</sub>	9.7			X		X
Thorite	ThSiO <sub>4</sub>	4.5-5.4			X		X
Topaz	Al <sub>2</sub> SiO <sub>4</sub> (F,OH) <sub>2</sub>	3.5-3.6			X		X
Tourmaline	(Na,Ca)(Mg,Fe <sup>2+</sup> ,Fe <sup>3+</sup> ,Al,Li) <sub>2</sub> Al <sub>6</sub> (BO <sub>3</sub> ) <sub>2</sub> Si <sub>2</sub> O <sub>6</sub> (OH) <sub>4</sub>	2.9-3.2		X →	(1)	(1,2) ←	X
Uraninite	UO <sub>2</sub>	11		X			X
Vermiculite	Mg <sub>2</sub> [Al,Si <sup>4+</sup> O <sub>10</sub> ](OH) <sub>2</sub> •nH <sub>2</sub> O	2.4-2.7			X		X
Wolframite	(Fe,Mn)WO <sub>4</sub>	6.7-7.5		X		X	
Wollastonite	CaSiO <sub>3</sub>	2.8-2.9			X		X
Wulfenite	PbMoO <sub>4</sub>	6.7-7.0			X	X	
Xenotime	YPO <sub>4</sub>	4.4-5.1		X			X
Zeolite	Hydrous Alumino-Silicate usually of Ca and Na	2.0-2.5			X		X
Zincite	ZnO	5.7			X	(1)	X
Zircon	ZrSiO <sub>4</sub>	4.7			X		X

### Explanation of Table:

Magnetic response is classified as (1) ferromagnetic: attraction to a conventional permanent magnetic separator at a metric flux density of less than 0.2 Tesla (2,000 gauss), (2) paramagnetic: attraction to the magnetic zone of a high intensity magnetic separator operating at greater than 0.2 to 2.0 Tesla (2,000-20,000 gauss), and (3) non-magnetic or diamagnetic: not-affected or repelled respectively from the magnetic force described in (2). Electrostatic response is classified as conductive or non-conductive based on the surface resistivity a mineral. Generally, minerals with a surface resistivity of less than 10<sup>11</sup> ohms will report as "conductors" using standardized settings on an electrostatic separator. Minerals with a surface resistivity above 10<sup>12</sup> ohms will report as "non-conductors" under similar equipment settings.

Notes: (1) Variations in response to magnetic and electrostatic separation will occur based on the actual composition of the mineral from varying locations and the presence of inclusions, (2) The response of some minerals can be varied by treatment at elevated temperature, i.e. rutile becomes more conductive at temperatures greater than 200 C, (3) Triboelectric (contact) charging can be used to selectively charge certain minerals; and they, in turn, can be attracted to an electrode of the opposite polarity, i.e. quartz from collophanite (pebble phosphate), (4) The separation of particles with a pronounced shape factor (mica) can also be accomplished by the use of a non-uniform electric field (dielectrophoresis).